

Production of a functionalized film for non-food applications from casein-based powders obtained from expired pasteurized milk

Marta SINDACO^{1*}, Thom HUPPERTZ^{2,3}, Hooriyeh MOHAMMADPOUR¹, Veronica ROSI¹, Sreedevi NALLITHODI¹, Luisa PELLEGRINO¹, Paolo D'INCECCO¹

¹ Università degli Studi di Milano, Dipartimento di Scienze per gli Alimenti, la Nutrizione e l'Ambiente, via Celoria 2, 20133 Milano.

² FrieslandCampina, Stationsplein 4, 3818 LE, Amersfoort, Netherlands

³ School of Food and Nutritional Sciences, University College Cork, Cork, T12 Y337, Ireland

* Correspondence: marta.sindaco@unimi.it

BACKGROUND

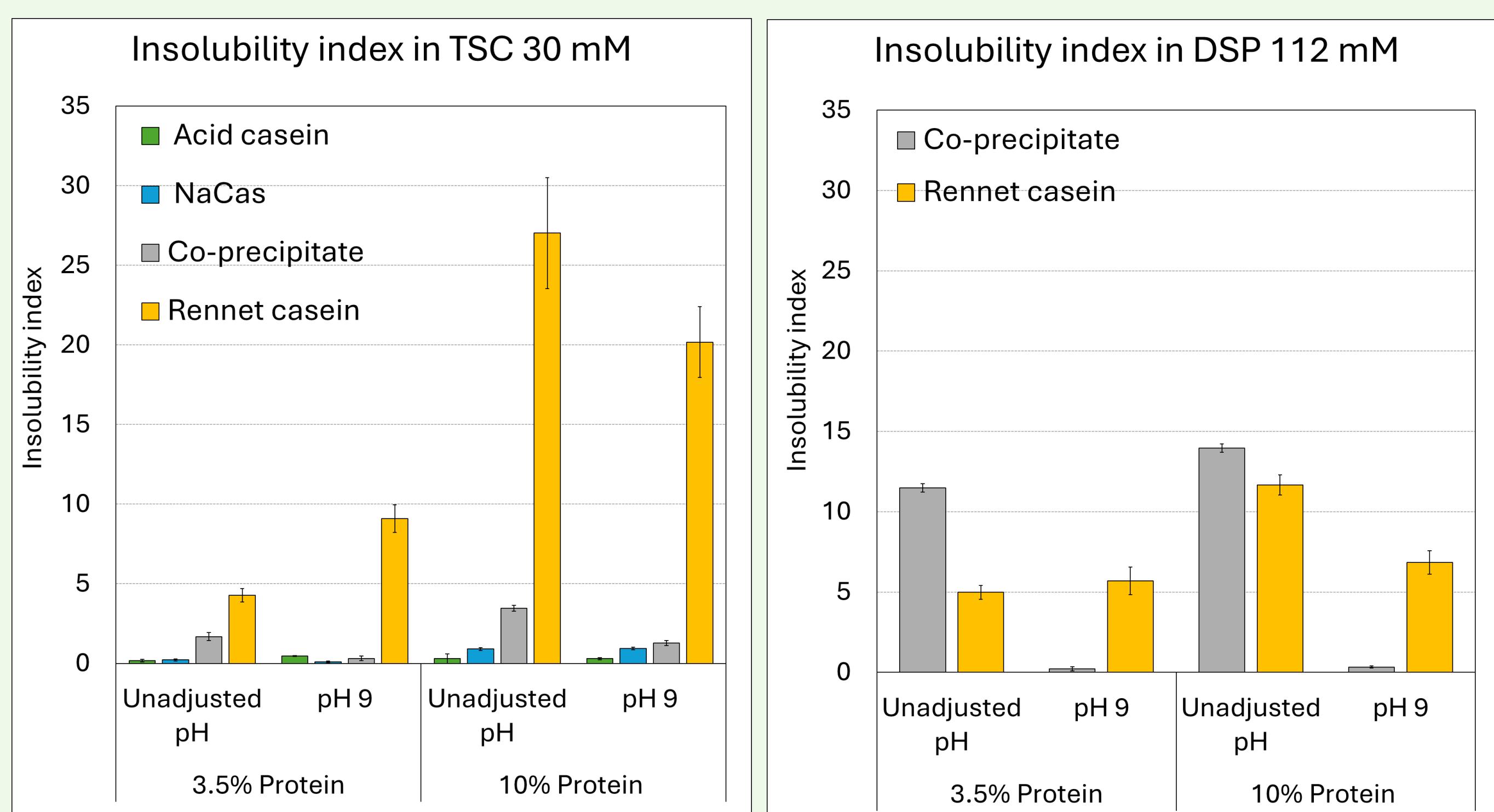
Pasteurized milk has a short shelf life, and one past its **expiry date**, it is classified as a special category 3 waste (Reg. CE 1069/2009). However, it still contains about 26 g/L of casein that can be recovered for non-food uses, such as packaging films [1]. **Casein** can be obtained as acid casein, rennet casein, co-precipitate or caseinates. Some of these powders often show **poor solubility** and rehydration, which can be improved by adjusting pH, temperature, or by adding calcium chelator [2]. Among them, sodium caseinate (NaCas) is water-soluble and suitable as film-forming biopolymers. **NaCas** films offer good barriers to non-polar substances but are hydrophilic and brittle, often requiring plasticizers or reinforcement. Protein crosslinking before film formation may improve strength of the three-dimensional network and modify functional properties. In this study, two reagent-free chemical approaches were tested to promote low-level **casein crosslinking** in NaCas film-forming solutions: (i) photo-oxidation leading to di-tyrosine (di-Tyr) formation [3]; (ii) β -elimination reactions leading to lysinoalanine (LAL) formation [4].

RESULTS and DISCUSSION

CASEIN-BASED POWDERS SOLUBILITY ASSESSMENT

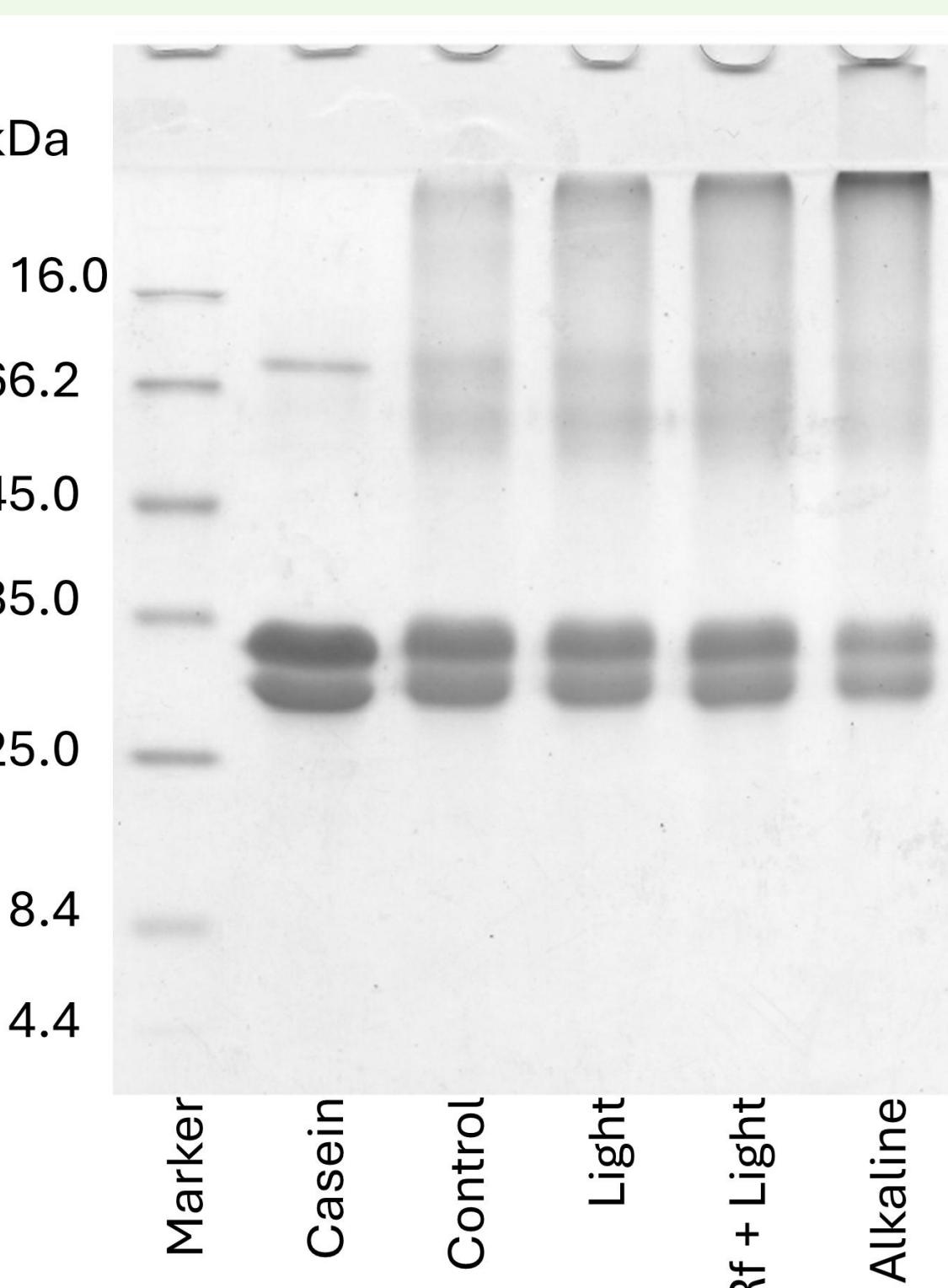
► The solubility of the powders was significantly affected by both pH and powder type:

- Acid casein and sodium caseinate: the insolubility index was always <1%, regardless of pH and reconstituted protein concentration;
- Co-precipitate: the solubility was strongly influenced by pH, with lower insolubility at alkaline pH;
- Rennet casein: showed the highest insolubility index in TSC among powders. The insolubility index at 10% protein improved in DSP vs TSC.



NaCas FILM-FORMING SOLUTIONS CHARACTERIZATION

NaCas solution treatment	LAL formation ($\mu\text{mol/g protein}$)	di-Tyr formation ($\mu\text{mol/g protein}$)
Control	3.88 \pm 0.1	0 \pm 0
Light	4.06 \pm 0.1	0.01 \pm 0
Rf + Light	3.46 \pm 0.05	1.39 \pm 0.02
Alkaline pH	15.86 \pm 0.3	0 \pm 0



Cross-link formation.

All NaCas solutions contained LAL, as all samples were subjected to the heat treatment. At alkaline pH, LAL was \sim 4 times higher than at unadjusted pH.

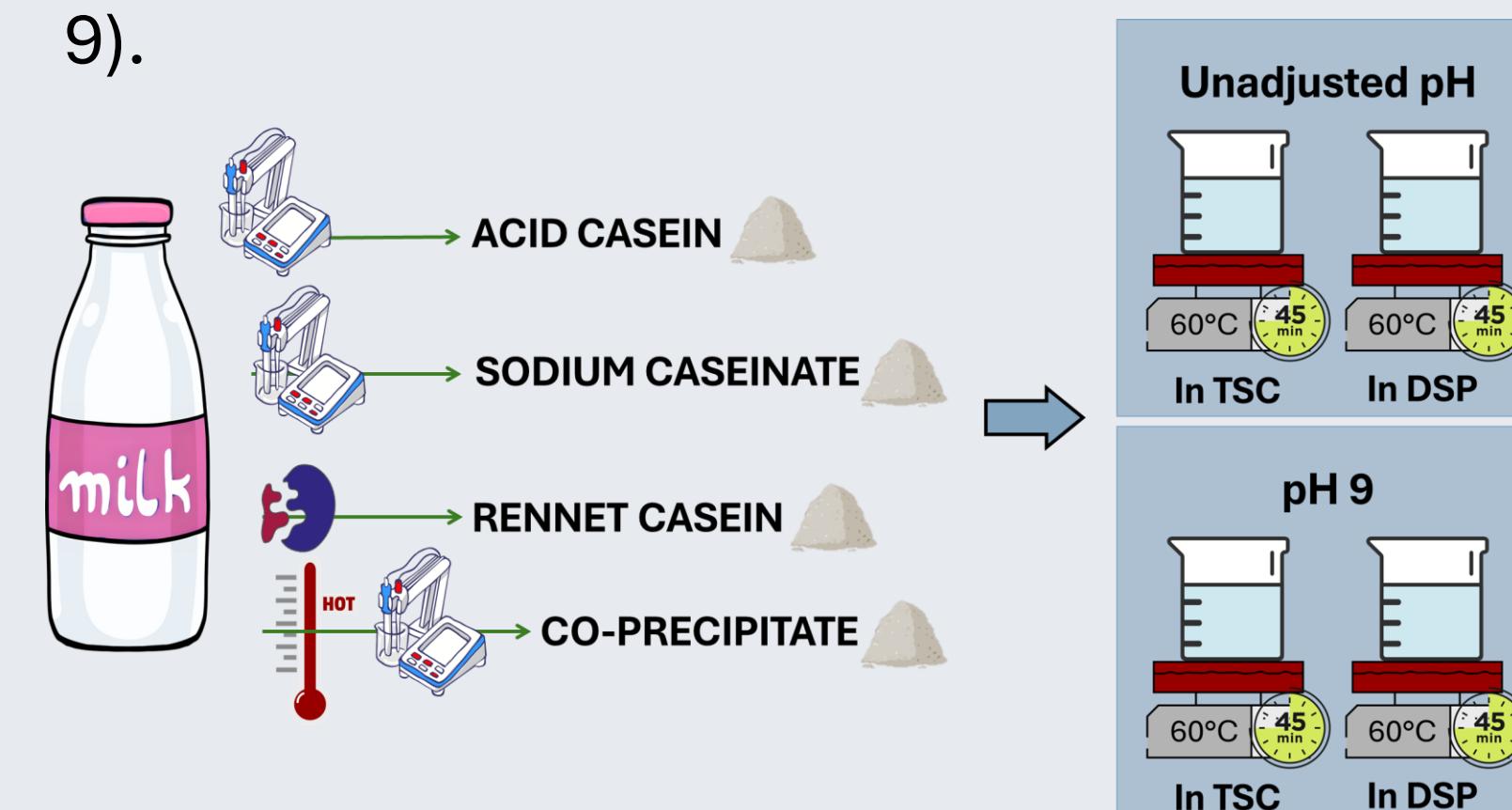
di-Tyr was absent in control and Alkaline pH samples, while light exposure alone induced only trace amounts. The addition of riboflavin increased di-Tyr formation confirming its role as a photo-sensitizer.

Protein aggregates formation.

Control NaCas solution already showed aggregates. Alkaline-thermal treatment further enhanced protein aggregation, with a strong decrease in monomers and accumulation of high-molecular-weight aggregates.

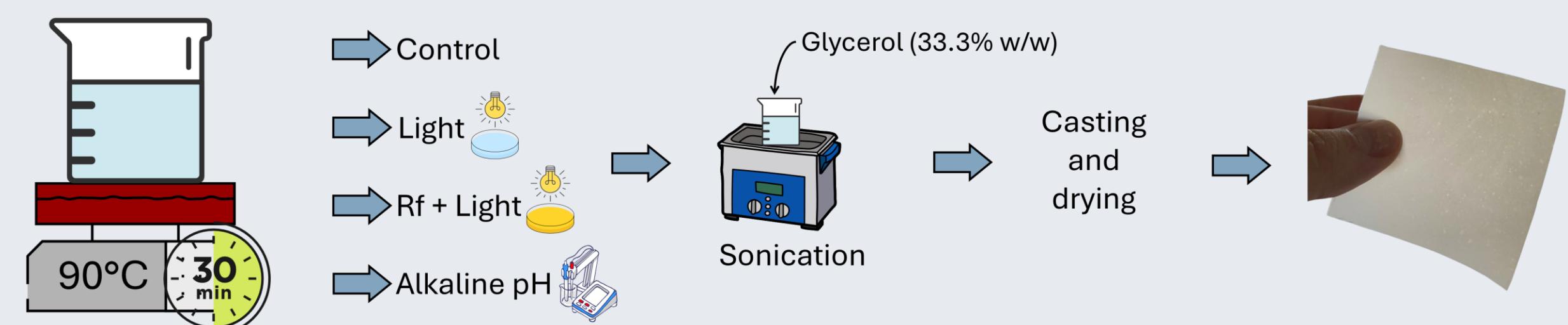
The contribution of di-Tyr to aggregation was minimal, as its levels were much lower than LAL and did not lead to additional aggregate formation.

► Recovery of casein-based powders from expired pasteurized semi-skimmed milk and solubilization of the powders through heat treatment at 60 °C for 45 minutes in 30 mM trisodium citrate (TSC) and 112 mM disodium phosphate (DSP). Powders were reconstituted at 3.5% and 10% protein, under two pH conditions (unadjusted and pH 9).



► **Determination of insolubility index.** Calculated as % (w/w) ratio between the pellet weight obtained after centrifugation (36 g for 10 min) and the weight of 1.5 mL of sample before centrifugation.

► NaCas (10% protein in MilliQ water) was used to prepare film-forming solutions under four conditions: (i) heating at 90 °C for 30 min (Control); (ii) exposition to LED light for 60 min then heating (90°C for 30 min) (Light); (iii) riboflavin addition followed by LED illumination (60 min) and heating (90°C for 30 min) (Rf + Light); (iv) pH adjustment to 9 with 1 M NaOH and heating (90°C for 30 min) (Alkaline pH).



► **Determination of LAL and di-Tyr formation** in film-forming solutions by LC-MS and HPLC-FL, respectively.

► **Determination of protein aggregates** in film-forming solutions by reducing SDS-PAGE.

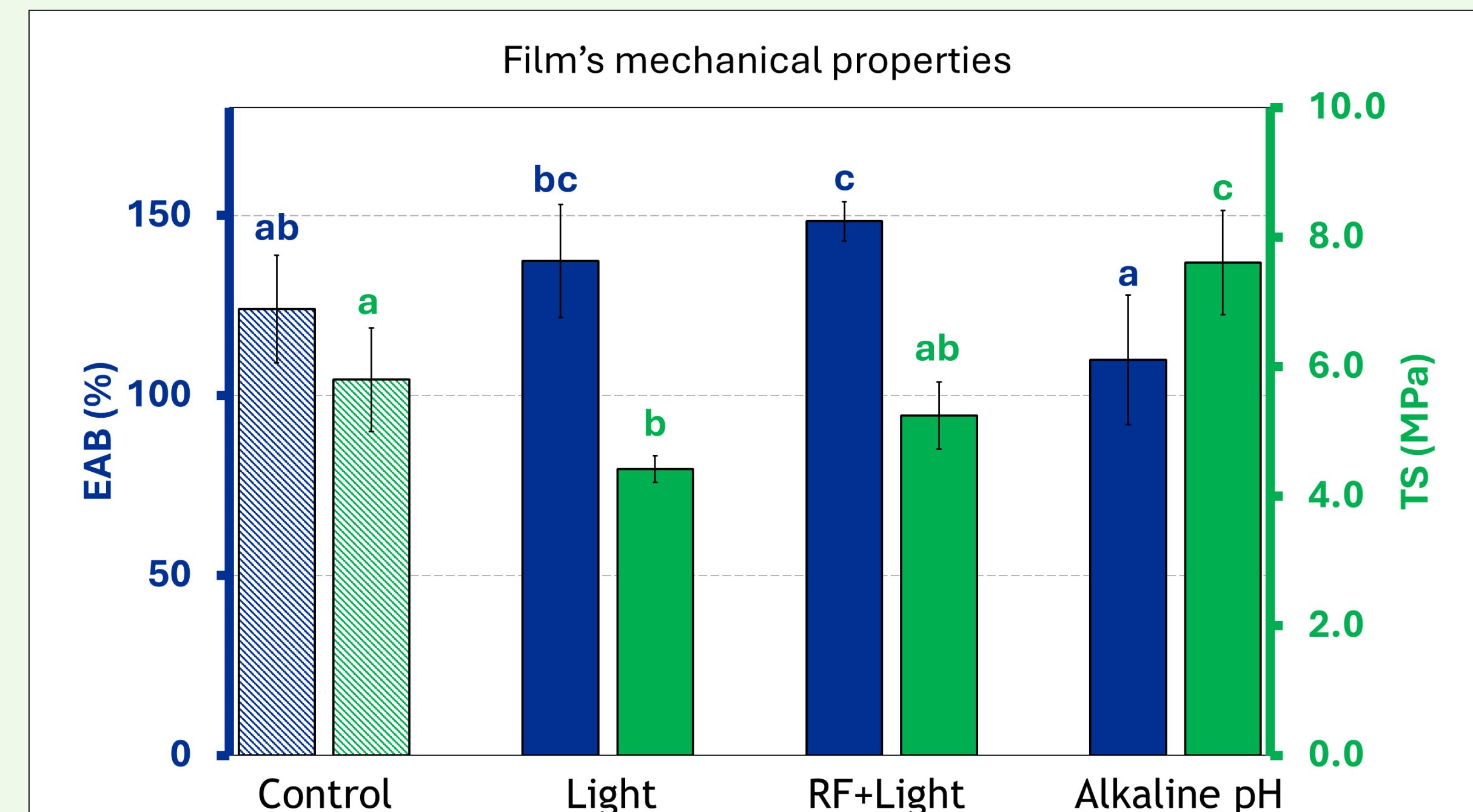
► **Mechanical properties** characterization of films.

CHARACTERIZATION OF NaCas FILMS

Mechanical properties.

Films derived from alkaline-thermal treated casein resulted in the highest tensile strength (TS) and the lowest elongation at break (EAB), consistent with the presence of extensive LAL-mediated crosslinking.

Films obtained from light-treated casein showed reduced tensile strength and increased elongation at break despite the minimal di-Tyr formation. The addition of riboflavin slightly improved both properties, though the effect remained limited.



CONCLUSION

Packaging prototype using Na-cas film:



This study confirmed that casein type is a key factor for its solubilization. Acid casein and sodium caseinate were highly soluble, while co-precipitate dissolved only at alkaline pH. Rennet casein remained poorly soluble except at 3.5% protein in TSC or 10% protein in DSP. Only NaCas films were functionalized using two green strategies, which enhanced their strength and elasticity, highlighting their potential for sustainable packaging films.

REFERENCES: (1) Gerna et al. "Strategies for exploiting milk protein properties in making films and coatings for food packaging: A review." *Foods* 12.6 (2023): 1271. (2) Crowley, et al. "Rehydration characteristics of milk protein concentrate powders." *Journal of Food Engineering* 149 (2015): 105-113. (3) Fuentes-Lemus et al. "α- and β-casein aggregation induced by riboflavin-sensitized photo-oxidation occurs via di-tyrosine cross-links and is oxygen concentration dependent." *Food chemistry* 256 (2018): 119-128. (4) Cattaneo, Masotti, Pellegrino. "Chemical modifications of casein occurring during industrial manufacturing of milk protein powders." *European food research and technology* 235.2 (2012): 315-323.